D. Bhagawan¹, P. Saritha², G. Shankaraiah¹, V. Himabindu^{1*}

FLUORIDE REMOVAL FROM GROUNDWATER USING HYBRID CYLINDRICAL ELECTROCOAGULATION REACTOR

¹Centre for Environment, IST, JNTUH, Kukatpally, Telangana, India; ²Civil Department, Malla Reddy Engineering College, Hyderabad, India *drvhimabindu@gmail.com, naik.bhagawan@gmail.com

Long-term consumption of water containing excessive fluoride can lead to fluorosis of the teeth and bones. Electrocoagulation is an electrochemical technique, in which a variety of unwanted dissolved particles and suspended matter can be effectively removed from an aqueous solution by electrolysis. Semi-continuous flow experiments for fluoride removal were undertaken to investigate the effects of the different parameters such as: applied voltage $(10-20\ V)$, flow rate $(150-450\ mL/min)$, initial pH (6-8), and initial fluoride concentration $(2-10\ mg/L)$ at lowest cost with novel reactor. The maximum of 8 mg/L fluoride was treated up to the World Health Organization drinking limits within 30 min residual time at a flow rate of 300 mL/min, with an applied voltage of 15 V at influent pH 7. The results obtained, showed that this novel semi-continuous flow electrocoagulation reactor is efficient for defluoridation of ground water supplies at lower cost.

Keywords: defluoridation, coagulant, ground water, flocs, sludge.

Introduction

Fluoride is an essential element to the human body. But at higher levels above than 1.5 mg/L consumption causes fluorosis. The main source of fluoride intake to the human being is drinking water. Groundwater is an important source of drinking water in widely ranged arid and semi-arid area [1]. Fluoride pollution in environment occurs through two different channels: natural sources and anthropogenic sources. Geochemical reactions in sub-surface and erosion of fluoride bearing rocks contribute fluoride to ground water. The discharge of industrial wastewater, such as semiconductor industries, aluminum industries and glass manufacturing industries, also contributes fluoride into water reserves, especially to groundwater [2]. Because of this the concentration

© D. Bhagawan, P. Saritha, G. Shankaraiah, V. Himabindu, 2019

- [11] Ichrak, Ayed Halleb, Nafaa Adhoum, Lotfi Monser // Ibid. 2013. 107. P. 150—157.
- [12] Anissa Aouni, Cheïma Fersi, Mourad Ben Sik Ali, Mahmoud Dhahbi // J. Hazard. Mater. 2009. **164**. P. 868—874.
- [13] Kollara V.R., Nad K., Mikelic I.L., Gustek S.F. // J. Environ. Sci. and Health. 2013. 48. P. 1543—1547.
- [14] *Sardari S.E., Javadian H.R., Katal R., Vafaie M.* // Korean J. Chem. Eng. 2013. **30**. P. 634–641.
- [15] *American* Public Health Association (APHA). Standard Methods for Examination of the Water and Waste water. [21st addition]. 2005.
- [16] *Chih, Wei-Lung Chou, Yi-Ming Kuo //* J. Hazard. Mater. 2009. **164**. P. 81–86.
- [17] Essadkia A.H., Gouricha B., Ch. Vial b. et al. // Ibid. 2009. **164**. P. 1325—1333.
- [18] *Emamjomeh M.M., Sivakumar M. //* J. Environ. Manag. 2009. **90**. P. 1204—1212.
- [19] *Bhagawan D., Saritha Poodari, Tulasiram Pothuraju et al.* // Environ. Sci. Pollut. Res. 2014. **21**. P. 14166–173.
- [20] *Feng Shen, Xueming Chen, Ping Gao, Guohua Chen //* Chem. Eng. Sci. 2003. **58**. P. 987–993.
- [21] *Behbahani M., Alavi Moghaddam M.R., Arami M.* // Desalination. 2010. **271**. P. 209—218.

Received 17.10.2016 Revised 23.05. 2017 Accepted 21.03. 2019